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Abstract	<p>Although membrane proteins account for 20–30% of the coding regions of all sequenced genomes and play crucial roles in many fundamental cell processes, there are relatively few membrane proteins with known 3D structure. This is likely due to technical challenges associated with membrane protein extraction, solubilisation, and purification. Membrane proteins are classified based on the level of interaction with membrane lipid bilayers, with peripheral membrane proteins associating non-covalently with the membrane, and integral membrane proteins associating more strongly by means of hydrophobic interactions. Generally speaking, peripheral membrane proteins can be purified by milder techniques than integral membrane proteins, whose extraction requires phospholipid bilayer disruption by detergents. Here, important criteria for strategies of membrane protein purification are addressed, with a focus on the initial stages of membrane protein solubilisation, where problems are most frequently encountered. Protocols are outlined for the successful extraction of peripheral membrane proteins, solubilisation of integral membrane proteins, and detergent removal which is important not only for retaining native protein stability and biological functions, but also for the efficiency of later purification techniques.</p>	
Key words (separated by '-')	Peripheral membrane protein - Integral membrane protein - Detergent - Protein purification - Protein solubilisation	

**Strategies for the Purification of Membrane Proteins** 2

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**Abstract** 4

Although membrane proteins account for 20–30% of the coding regions of all sequenced genomes and play crucial roles in many fundamental cell processes, there are relatively few membranes with known 3D structure. This is likely due to technical challenges associated with membrane protein extraction, solubilisation, and purification. Membrane proteins are classified based on the level of interaction with membrane lipid bilayers, with peripheral membrane proteins associating non-covalently with the membrane, and integral membrane proteins associating more strongly by means of hydrophobic interactions. Generally speaking, peripheral membrane proteins can be purified by milder techniques than integral membrane proteins, whose extraction requires phospholipid bilayer disruption by detergents. Here, important criteria for strategies of membrane protein purification are addressed, with a focus on the initial stages of membrane protein solubilisation, where problems are most frequently encountered. Protocols are outlined for the successful extraction of peripheral membrane proteins, solubilisation of integral membrane proteins, and detergent removal which is important not only for retaining native protein stability and biological functions, but also for the efficiency of later purification techniques.

**Key words:** Peripheral membrane protein, Integral membrane protein, Detergent, Protein purification, Protein solubilisation 18  
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**1. Introduction** 20

Membrane proteins are associated with the membrane of a cell or particular organelle and are generally more problematic to purify than water-soluble proteins. Membrane proteins represent approximately 20–30% of the open-reading frames of an organism's genome (1) and play crucial roles in basic cell functions including signal transduction, energy production, nutrient uptake, and cell–cell communication. Additionally, many drugs and drug candidates target membrane proteins (2, 3). However, less than 2% of the listed 3D structures in the protein data bank (4) are 21  
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30 membrane proteins, a fact that is likely due to the technical challenges  
31 associated with membrane protein solubilisation and purification  
32 in sufficient quantities for crystallisation (5).

33 Membrane proteins are classified into peripheral and integral  
34 membrane proteins, which are associated to varying degrees with  
35 the phospholipid bilayer (6, 7). Peripheral or extrinsic membrane  
36 proteins interact with the membrane surface non-covalently by  
37 means of electrostatic and hydrogen bonds. Peripheral membrane  
38 proteins can be recruited to the membrane during signalling  
39 events or are constitutively localised to the membrane. Integral or  
40 intrinsic membrane proteins are more strongly associated with  
41 the membrane and interact with hydrophobic moieties of the  
42 phospholipid bilayer. They contain one or more characteristic  
43 runs of apolar amino acids that span the lipid bilayer (6). Integral  
44 membrane proteins are further classified into Type I, which are  
45 positioned so that their COOH-terminus is embedded in the  
46 cytosol or Type II, which are positioned with the NH<sub>2</sub>-terminus  
47 in the cytosol. Although there is no single protocol for the purifi-  
48 cation of membrane proteins, it is the initial stages of membrane  
49 protein solubilisation where problems are most frequently  
50 encountered. This chapter addresses important criteria for mem-  
51 brane protein extraction and solubilisation. The Subheading 3  
52 describes protocols for the successful extraction of peripheral  
53 membrane proteins, solubilisation of integral membrane proteins,  
54 and detergent removal which is important not only for retaining  
55 native protein stability and biological functions, but also for the  
56 efficiency of later purification techniques.

### 57 **1.1. Considerations** 58 **for Membrane Protein** 59 **Purification**

60 The analysis of membrane proteins represents a significant techni-  
61 cal challenge in the field of proteomics and there are several rea-  
62 sons why the purification of membrane proteins is more difficult  
63 than that of water-soluble proteins. Firstly, endogenous expres-  
64 sion of membrane proteins is quite low and usually quite large  
65 quantities of protein are required for structural investigations.  
66 Additionally, integral membrane proteins are extremely hydro-  
67 phobic and often require high concentrations of detergents for  
68 solubilisation. Membrane proteins have the tendency to form  
69 aggregates, even in the presence of detergents, resulting in the  
70 reduction of efficiency of subsequent separation techniques (8).  
71 The choice of detergent may also affect the efficiency of down-  
72 stream protein purification procedures. For example, ion-exchange  
73 chromatography (see Chapter 12) should not be carried out in  
74 the presence of charged detergents, and hydrophobic interaction  
75 chromatography (see Chapter 24) can be problematic in the pres-  
76 ence of all detergents (8). In such cases detergents can be removed  
(see Subheading 1.4). Once solubilised, the membrane proteins  
are often more susceptible to degradation by proteases. Thus,  
addition of protease inhibitors such as ethylenediamine tetraacetic

acid (EDTA), which inactivates metalloproteases, or phenylmethyl sulfonyl fluoride (PMSF), which inhibits serine proteases, needs to be considered.

It is also worthwhile considering the availability of efficient functional assays to detect the protein of interest at different stages during the purification process, for example, measurement of enzymatic activity and immuno- or ligand-binding assays. Given the unique properties of individual proteins, it is usually necessary to determine appropriate assays on a case-by-case basis (8). There is no single protocol for obtaining membrane protein purification; more likely a series of methods are needed, depending on the particular needs of the investigator.

### 1.2. Peripheral Membrane Protein Extraction

Peripheral membrane proteins can be dissociated using relatively mild techniques that break the electrostatic or hydrogen bonds between the peripheral proteins and the membrane, without total membrane disruption. Common dissociating reagents for the extraction of peripheral membrane proteins are listed in Table 1. Extractions using buffers containing high salts are useful as they decrease electrostatic interactions between proteins and charged lipids (6). Chaotropic ions disrupt hydrophobic bonds present in the membrane surface and promote the transfer of hydrophobic groups from non-polar environment to the aqueous phase (6). Usually, extraction procedures employing high ionic strength NaCl and KCl, alkaline or acidic buffers, and metal chelators result in a relatively distinct separation between solubilised peripheral proteins and membrane-associated integral membrane proteins (7). High pH causes the fractionation of peripheral membrane proteins from integral membrane proteins by disrupting sealed membrane structures without denaturing the lipid

**Table 1**  
**Treatments for the extraction of peripheral membrane proteins**

Treatment type	Example
Acidic buffers	pH 3.0–5.0
Alkaline buffers	pH 8.0–12.0 (e.g. 100 mM Na <sub>2</sub> CO <sub>3</sub> , pH 11.3, see Subheading 3.1)
Chaotropic ions	I <sup>-</sup> , ClO <sub>4</sub> <sup>-</sup> , SCN <sup>-</sup>
Denaturing agents	8 M urea or 6 M guanidine hydrochloride
Metal chelators	10 mM EDTA or EGTA
Salt solutions/high ionic strength	1 M NaCl or KCl

106 bilayer or extracting integral membrane proteins (9). The high  
107 pH method for extraction of peripheral membrane proteins is  
108 described in Subheading 3.1 of this chapter. It is worthwhile  
109 determining the effect of the buffer on any enzymatic activity the  
110 protein of interest may have, and potential interactions the buffer  
111 may have with any column matrix that will be used at later stages  
112 in the purification process. Additionally, buffer cost may need to  
113 be considered if large-scale preparations are to be carried out.

114 Following extraction (i.e. breaking of electrostatic and H  
115 bonds between peripheral protein and the membrane) in the cho-  
116 sen buffer for 10–30 min, the remaining membrane bilayer and  
117 its associated integral proteins are separated by centrifugation  
118 (30–60 min,  $100,000 \times g$ ) and the released peripheral membrane  
119 proteins are recovered in the supernatant (7, 10).

### 120 **1.3. Integral** 121 **Membrane Protein** 122 **Extraction**

123 In order to solubilise integral membrane proteins, it is necessary  
124 to disrupt the lipid bilayer, which may be achieved with organic  
125 solvents, but is more commonly accomplished using detergents.  
126 Extraction using the organic solvent N-butanol (see  
127 Subheading 3.2) uses a biphasic system for solubilising proteins  
128 from membranes into dilute aqueous buffers. The low solubility  
129 of N-butanol in water, combined with its lipophilicity, minimally  
130 denatures proteins (9). Detergents are amphipathic molecules  
131 that contain both hydrophobic and hydrophilic moieties and form  
132 micelles in water. A micelle is a cluster of detergent molecules in  
133 which the hydrophilic head moieties face outward. Detergents  
134 solubilise proteins by binding to the hydrophobic parts of the  
135 protein on one side and interacting with the aqueous parts on the  
136 other side (8). The detergent of choice should sufficiently solubi-  
137 lise the membrane protein without irreversibly denaturing it.  
138 Detergents can be ionic, non-ionic, or zwitterionic. A list of com-  
139 monly used detergents for extraction of integral membrane pro-  
140 teins is shown in Table 2. Selection of a particular detergent  
141 depends on the properties of the protein of interest and the given  
142 aims of subsequent experiments involving the purified protein. If  
143 there is little information in the literature on the purification of  
144 similar proteins, or if one is purifying a particular protein for the  
145 first time, it is often necessary to screen a number of detergents in  
146 order to optimise protein solubilisation. Membrane aliquots  
147 should be incubated with various concentrations of commonly  
148 used detergents and incubation time, buffer concentration, salt  
149 solutions, and temperature conditions necessary for optimal solu-  
150 bilisation should be determined.

151 When screening potential detergents, it is important to be  
152 aware of the unique critical micelle concentration (CMC), which  
is the concentration of free detergent at which the transition from  
disperse detergent molecules to a micellar structure occurs (10).  
Since solubilisation corresponds to the removal of the protein

**Table 2**  
**Detergents used for extracting integral membrane proteins**

Detergent type	Name	Alternative chemical name	CMC (mM)
Ionic	CTAB	Cetyltrimethylammonium bromide	1.0
	Sodium cholate		~10
	Sodium deoxycholate		~2
Non-ionic	Big Chap	<i>N,N</i> -bis(3-D-gluconamidopropyl) cholamide	3.4
	C <sub>12</sub> E <sub>8</sub>	Octaethylene glycol monododecyl ether	<0.1
	Triton X-100	Nonaethylene glycol octylphenol ether	0.3
Zwitterionic	CHAPS	3-[(3-cholamidopropyl) dimethylammonio]-1-propanesulfonate	3–10
	CHAPSO	3-[(3-cholamidopropyl) dimethylammonio]-2-hydroxypropane-1-sulfonate	4–8
	LDAO	Dodecyltrimethylamine oxide	~1

CMC critical micelle concentration

from the membrane into the detergent micelle, the CMC is the minimal concentration of detergent necessary to form the required micellar structure for protein extraction (10). CMC values, some of which are listed in Table 2, vary between different detergents, but are usually available from the detergent manufacturer.

Additional considerations when choosing detergents include evaluating the effects of a given detergent on the structural and functional properties of the protein of interest. The effects of detergents on the protein stability may be checked during preliminary screens using different detergents. The compatibility of the chosen detergent with subsequent purification steps should also be considered as certain detergents may affect the efficiency of certain chromatographic techniques. For example, charged detergents may cause problems using assays based on charge difference, such as ion-exchange chromatography (see Chapter 2), and lectin chromatography which may be used to affinity purify subsets of glycoproteins is especially sensitive to high concentrations of a variety of detergents (7, 8). It is often necessary to remove or replace detergents to overcome these problems, thus the ease at which excess detergent can be removed from the solubilised protein fraction should be considered (see Subheading 1.4).

When solubilising integral membrane proteins, buffered stock solutions at a physiological pH environment should be prepared containing the membrane preparation, detergent, and protease inhibitors, such as EDTA, EGTA, and/or PMSF (10). Membrane preparations are used at a final protein concentration of 1–5 mg/mL

179 and are solubilised by detergent concentrations of 0.1–5% (v/v)  
 180 (7, 10). The mixture should be stirred gently for 1 h at room tem-  
 181 perature or 4°C, followed by centrifugation for 1 h at 100,000 × *g*  
 182 at 4°C. Generally speaking, retention of a membrane protein in the  
 183 supernatant following centrifugation for 60 min at 100,000 × *g* after  
 184 solubilisation defines this protein as soluble (7). The pellet may sub-  
 185 sequently be washed to remove residual detergent and finally re-  
 186 suspended in the appropriate buffer (10). Protein recovery and activity  
 187 should be investigated in both the pellet and supernatant at this  
 188 stage. The procedure for solubilising membrane proteins using the  
 189 non-ionic detergent Triton X-100 is outlined in Subheading 3.3.

190 **1.4. Removal**  
 191 **of Detergents**  
 192 **from Membrane**  
 193 **Protein Fractions**

The high detergent concentrations that are often required during  
 the initial extraction of integral membrane proteins could poten-  
 tially affect the stability and subsequent analysis of the isolated  
 membrane proteins; therefore, excess detergent should be  
 removed or exchanged for an alternative detergent prior to sub-  
 sequent purification procedures. Examples of methods used to  
 remove or exchange detergents are listed in Table 3. The choice  
 of technique depends on the unique properties of the detergent  
 used and the concentration range of the protein fraction.

Successful detergent exchange or removal can be achieved  
 using various chromatographic supports, followed by extensive  
 washing with the desired buffer, containing the new detergent if  
 necessary (6). Alternatively dialysis can be carried out to facilitate  
 detergent exchange or removal. The efficiency of dialysis depends  
 on the CMC and micelle molecular weight, which is determined  
 by the aggregation number of detergent molecules (11).  
 Most detergents with linear alkyl hydrophobic groups (e.g. Triton

t3.1 **Table 3**  
 t3.2 **Commonly used techniques for detergent removal/**  
 t3.3 **exchange**

t3.4	<b>Technique</b>	<b>Reagent</b>
t3.5	Affinity chromatography	Ligand immobilised sepharose
t3.6	Equilibrium dialysis	Appropriate buffer or water
t3.7	Gel permeation chromatography	Sephadex G-25 (GE Healthcare)
t3.8	Hydrophobic interaction	Bio-Beads SM-2 (Bio-Rad)
t3.9	chromatography	
t3.10	Ion-exchange chromatography	Dowex 1-X2 (Sigma-Aldrich)
t3.11	Precipitation	Acetone
t3.12	Ultrafiltration	High molecular weight cut-off
t3.13		membrane

X-100) have a high micelle molecular weight value and do not pass through dialysis membranes (6). Detergents with a low micelle molecular weight and high CMC (e.g. bile acids and their derivatives) can be removed by dialysis (6). A protocol for dialysis is described in Subheading 3.4 of this chapter. Detergent removal by means of chromatographic supports (see Subheading 3.5) is relatively work-intensive, but is a more rapid procedure than dialysis so can be advantageous in cases where protein stability is an issue.

Following initial extraction of membrane proteins, solubilisation using detergent, and detergent removal or exchange, membrane proteins can then be purified to homogeneity using a variety of protein purification techniques, depending on the particular needs of the investigator and the given properties and abundance of the protein of interest. Because there is no single procedure to characterise membrane proteins, the key importance of membrane purification lies with the initial extraction and solubilisation steps, in order to generate a high yield of pure protein in its native biologically active state. The Subheading 3 of this chapter outlines examples of techniques used during the important initial stages of membrane protein purification.

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## 2. Materials

### 2.1. Fractionation of Peripheral and Integral Membrane Proteins Using High pH

1. High pH buffer: 100 mM Na<sub>2</sub>CO<sub>3</sub>, pH 11.3. 228
2. Dounce homogeniser, e.g. Potter-Elvehjem PTFE pestle and glass tube (Sigma-Aldrich). 229  
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3. Ultracentrifuge, e.g. Thermo Scientific Sorvall WX. 231

### 2.2. Extraction of Membrane Proteins Using Butanol

1. N-butanol. 232
2. Cooled bench top centrifuge, e.g. Eppendorf centrifuge 5417R. 233  
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### 2.3. Extraction of Membrane Proteins Using Triton X-100

1. TE buffer: 10 mM Tris-HCl, 2 mM EDTA. 235
2. 2% Triton X-100 in phosphate buffered saline (PBS) (see Note 1). 236  
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3. Ultracentrifuge, e.g. Thermo Scientific Sorvall WX. 238

### 2.4. Removal of Non-ionic Detergents by Detergent-Adsorption Chromatography

1. Columns with a bed volume of approx. 5 mL (e.g. Econo-column, Bio-Rad). 239  
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2. Commercially available detergent absorption matrix (e.g. Bio-Beads SM-2, Bio-Rad, see Note 2). 241  
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3. Blocking buffer: 0.1% (w/v) bovine serum albumin in 50 mM Tris-HCl, pH 7.4, 0.15 M NaCl (see Note 3). 243  
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4. Washing buffer: 50 mM Tris-HCl, pH 7.4, 0.15 M NaCl. 245



246 **2.5. Removal of**  
247 **Detergent with Low**  
248 **Micelle Size and High**  
249 **CMC by Dialysis**

1. Dialysis tubing with a molecular weight cut-off of approx. 10,000 Da.
2. Wash buffer: 100 mM NaHCO<sub>3</sub>, 50 mM EDTA.
3. Dialysis buffer: 20 mM Tris-HCl, pH 7.4, 0.15 M NaCl.

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250 **3. Methods**

251 **3.1. Fractionation of**  
252 **Peripheral and Integral**  
253 **Membrane Proteins**  
254 **Using High pH**

1. Re-suspend the membrane fraction (see Note 4) at a concentration of <2 mg/mL in high pH buffer (see Notes 5 and 6).
2. Homogenise the suspension in a dounce homogeniser using six to eight strokes.
3. Maintain at 4°C for 30 min. Mix by vortexing three times during this period.
4. Pellet the membrane fraction by centrifugation for 60 min at 100,000 × *g* at 4°C and transfer the supernatant, which contains the peripheral membrane proteins, into a fresh tube and assay for protein (see Note 7).

261 **3.2. Extraction**  
262 **of Membrane Proteins**  
263 **Using Butanol**

1. Add an equal volume of N-butanol to the membrane fraction (see Note 4) and maintain at 4°C.
2. Centrifuge at 500 × *g* at 4°C for 10 min to separate the mixture into an upper phase containing butanol and membrane lipids and a lower aqueous phase containing solubilised proteins. Lipid rich material is localised to the interface.
3. Separate the upper and lower aqueous phases into separate tubes.
4. Dialyze the aqueous phase against a large volume of water or suitable buffer.
5. Assay the dialysed aqueous phase for protein (see Note 8).

272 **3.3. Extraction of**  
273 **Membrane Proteins**  
274 **Using Triton X-100**

1. Re-suspend cells in TE buffer at a concentration of 1 × 10<sup>7</sup> cells/mL.
2. Centrifuge the cells at 40,000 × *g* for 10 min. Remove the supernatant and add fresh TE.
3. Repeat this step and re-suspend the cells in approximately 1 mL of TE.
4. Add cells drop-wise to the 2% Triton X-100 while stirring (see Note 9).
5. Allow to solubilise for 30 min at 4°C.
6. Centrifuge at 100,000 × *g* for 30 min at 4°C.
7. Transfer the supernatant to a fresh tube and assay for protein (see Note 10).

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**3.4. Removal of  
Non-ionic Detergents  
by Detergent-  
Adsorption  
Chromatography**

1. Before starting, ensure that the protein fraction containing the non-ionic detergent (e.g. Triton X-100) has a concentration of >1 mg/mL (see Note 11) and that the molecular weight of the protein to be recovered is large enough to avoid entrapment in the pores of the affinity matrix. 284  
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2. Apply distilled water to the column matrix, followed by blocking buffer. Next, apply washing buffer to the column and repeat wash step. 289  
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3. Transfer the protein fraction to the column matrix (see Note 12). 292  
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4. Collect 0.5–1 mL fractions and assay for protein. 294

**3.5. Removal of  
Detergent with Low  
Micelle Size and High  
CMC by Dialysis**

1. Prepare the dialysis tubing by boiling a section in washing buffer for 10 min (see Note 13). Then boil the dialysis tubing in distilled water for 10 min, followed by washing thoroughly in distilled water. 296  
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2. Transfer the solubilised membrane protein fraction into the dialysis tubing (see Note 14) which is securely closed at one end by either tying a double-knot in the tubing or securing it with a plastic clamp (see Note 15). 299  
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3. Remove air bubbles and seal the dialysis tubing, allowing for a volume increase during dialysis. Check the integrity of the seal to ensure no leakage occurs. 303  
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4. Place the tubing in a beaker containing a large external volume (approx. 5 L) of the appropriate buffer. Dialyze with gentle stirring at 4°C. Change the external buffer regularly. 306  
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5. When the dialysis is finished, remove the dialysis tubing and wash the outside. Hold the tubing and carefully remove the upper clamp. Using a Pasteur pipette, transfer the protein fraction to a new tube (see Note 16). 309  
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**4. Notes**

1. Make a stock solution of 20% Triton X-100 by weighing 2 g Triton X-100 and adding PBS up to 10 mL and stirring gently until fully dissolved. Store the stock solution at 4°C. 314  
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2. Bio-Beads are macro-porous polystyrene beads and have a high surface area that adsorbs organics with a molecular weight of <2,000 from aqueous solution. They may be used to remove Triton X-100 from protein fractions. Due to the presence of linear alkyl hydrophobic groups, Triton X-100 has a high micelle molecular weight value and does not pass through dialysis membranes. Detergents with a low micelle molecular weight and high CMC (e.g. bile acids and their derivatives) can be removed by dialysis (see Subheading 3.5). 317  
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3. Bovine serum albumin is used as a bulk carrier protein to saturate non-specific protein binding sites and minimise protein loss during this procedure.
  4. The starting material depends on the source from which the membrane proteins are being purified. Membrane proteins can be successfully isolated from plant and animal tissues or cell cultures, bacteria, yeast, and fungi. Animal tissues can be broken easily with a mixer or blender. Due to the presence of robust cell walls, unicellular organisms like yeast or bacteria and plant cells are more difficult to disrupt. Different techniques for breaking down cell walls include glass bead milling, grinding mills, homogenization, ultrasonication, osmotic shock, repeat freeze thawing, and enzymatic lysis (8). If possible, the protein should be prepared from sources where it is in high abundance, as a certain amount of protein may be lost during the purification process. The starting material can be enriched if the target protein is known to be associated with the plasma membrane, mitochondria, or nucleus. During initial steps of membrane protein isolation, cytosolic proteins can be removed to obtain an enriched preparation of membranes containing the protein of interest. Soluble cytoplasmic proteins are extracted by cell disruption in a neutral pH, isotonic, and detergent-free buffer (7), followed by differential centrifugation or purification using sucrose gradient centrifugation.
  5. The pH of the working buffer should be tested following addition of any protease inhibitors, as addition of such components may alter the final pH of the buffer.
  6. It is worthwhile determining the effect of the high pH buffer on any enzymatic activity the protein of interest may have, and considering potential interactions the buffer may have with any column matrix that will be used at later stages in the purification process.
  7. A suitable protein concentration assay should be considered. Options include measuring ultra-violet absorbance at 280 nm, or using one of several commercially available dye-binding assays, such as the Bradford assay, the bicinchonic acid (BCA) assay, or other assays (see Chapter 13).
  8. It is worthwhile to keep the butanol phase for protein assays as it may contain extremely hydrophobic proteins that are difficult to solubilise.
  9. The effect of the Triton X-100 solubilisation procedure on the structural and functional properties of the protein of interest should be evaluated during preliminary screening experiments. In order to maintain catalytic activity, the membrane protein should be dissolved under optimal conditions for stability at a detergent/protein ratio that is not much above the

372 minimal detergent/protein ratio required for solubilisation (8).  
 373 Additionally, proteins are more susceptible to protease attack  
 374 following solubilisation with detergents, so protease inhibi-  
 375 tors are necessary to prevent protein degradation. Premixed  
 376 cocktails of commonly used protease inhibitors are now avail-  
 377 able commercially from a variety of companies including  
 378 Roche, Sigma-Aldrich, and Pierce. It is recommended to  
 379 carry out purification procedures at 4°C in order to minimise  
 380 proteolysis (see Chapter 4). Additionally, the effects of Triton  
 381 X-100 on subsequent purification techniques should be eval-  
 382 uated. Replenish protein stabilising additives or protease  
 383 inhibitors if they are removed or inactivated at any stage in  
 384 the experiment, for example EDTA is removed by hydroxy-  
 385 apatite chromatography (8). If possible, minimise any purifi-  
 386 cation steps that add new detergents or alter the original  
 387 detergent/lipid ratio.

- 388 10. Due to the presence of aromatic groups, Triton X-100 has  
 389 substantial UV absorbance at 280 nm, thus an alternative  
 390 protein concentration assay should be carried out. For the  
 391 same reason, Triton X-100 is not suitable for subsequent  
 392 purification steps involving column chromatography with UV  
 393 monitoring of the fractions. As an alternative, bile salts and  
 394 their derivatives including CHAPS and CHAPSO can be used  
 395 for solubilisation.
- 396 11. A high concentration is necessary to allow for any loss of pro-  
 397 tein during the procedure.
- 398 12. Use washing buffer to dissolve the protein fraction for opti-  
 399 mum detergent binding.
- 400 13. As dialysis tubing is susceptible to cellulolytic micro-organ-  
 401 isms, gloves should be worn when handling the tubing.
- 402 14. A small funnel may be used to aid transfer of the protein frac-  
 403 tion into the dialysis tubing.
- 404 15. Prior to transferring the protein fraction into the dialysis tubing,  
 405 the integrity of the membrane and clamp/knot can be tested by  
 406 applying water or buffer and checking the tubing for leaks.
- 407 16. Avoid losing dialyzed samples by carefully opening the tubing  
 408 over a larger glass beaker to collect any accidental spillage.

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Uncorrected Proof